Srikrishna College

(Affiliated to University of Kalyani)

Bagula, Nadia, West Bengal.

Course Outcome

The learning outcome-based curriculum framework for B.Sc. degree in Chemistry is fundamental approach of learning a broad framework of chemical science and its application to our world.

The graduate students in chemistry are being upskilled with different equipment and instrumental knowledge which would empower them in their higher education, research field and industrial interface. The curriculum induces critical thinking, basic psychology, scientific reasoning, moral ethical reasoning and lays emphasis on the objectivity and employability for the students.

- ❖ The course helps to acquire chemistry knowledge which help further for higher studies and research in this field as well as develop students' ability and skill to acquire expertise over solving both theoretical and applied chemistry problems.
- Students will have a firm foundation in the fundamentals and application of current chemical and scientific theories including those in Analytical, Inorganic, Organic and Physical Chemistry.
- ❖ Able to carry-over the course with a broad and thorough knowledge in chemistry with chemical concepts, principles, and theories.
- Students will be able to design and carry out scientific experiments as well as accurately record and analyse the results of such experiments and able to explain the synthesis and analysis.
- students will be able to understand the characterization of materials and understand the basic principle of equipment used in the chemistry laboratory.
- Students will be proficient in problem solving, critical thinking and analytical reasoning as applied to real life scientific problems.
- Course will enable them to explore new areas of research in both chemistry and allied fields of science and technology.
- Student are capable of expressing the subject through technical writing as well as through oral presentation

- Students will be a critical thinker and problem solver, Team player, Skilled project manager.
- ❖ Students will be empowered to explain why chemistry is an integral activity for addressing social, economic, and environmental problems.
- . Enhance the ability to use computers for chemical simulation and computation.

> Honours Course wise Credit Distribution and Outcome

Course	Total	Total	Credit					
	no of	Theor	Practi	Core	Critical	Analytical	Resea	Team
	Paper	у	cal	compete	Thinking	Reasoning	rch	Work
	S			nce			Skill	
Core	14	56	28	$\sqrt{}$	$\sqrt{}$	V	$\sqrt{}$	V
Courses								
Disciplin	4	16	8	V	V	V	$\sqrt{}$	V
e Specific								
Elective								
Generic	4	16	8	$\sqrt{}$	X	V	X	$\sqrt{}$
Elective								
Skill Enhance	2	4	-	V	V	X	X	X
ment								

CBCS 1st SEM (Honours)

Course code	Course title	Course Outcome	Credit
CHEMHT-1 (Inorganic +Physical)	i)Extra nuclear structure of atom and Periodic properties, ii) Kinetic Theory and Gaseous state, iii)Chemical Thermodynamic s - I	 i) To familiarised students about the basic concept of atomic structure and it's relation with the elements of modern periodic table. ii) To understand about concept of the Kinetic theory of gases and differentiate between real and ideal gases through different concepts. iii) To learn the concepts of 1st law of thermodynamics and related state and path functions of different chemical processes. 	4
СНЕМНР-1	Inorganic Chemistry – IA & Physical Chemistry - IA	i)To illustrate the acid-base reactions of the different binary mixtures and idea about different standard solution, ii)To give knowledge about exothermic and endo thermic reactions, iii)To provide Concept of pH using colour matching method.	2
СНЕМНТ-2	Theory: Basics of Organic Chemistry, Bonding and Physical Properties, General Treatment of Reaction Mechanism and Stereochemistry	i)To acute students with the fundamental of organic chemistry such as bonding, physical properties and rection mechanism. ii)To understand geometries, chirality, symmetry of organic molecules and it's relation to optical activity, relatives and absolute configuration of the organic molecules.	4

Chemistry – I binary mixture based upon solubility	
determination of boiling points of different	
organic compounds	
ii)To learn specific chemical tests to	
identify different organic compounds.	

CBCS 2nd SEM honours

Course code	Course title	Course Outcome	Credit
CHEMHT-3 (Inorganic +Physical)	i)Redox reactions and Precipitation reactions, Acid- Base Concepts and Solvents ii)Chemical Thermodynamic s – II, Chemical kinetics	i) To acquire the knowledge on oxidation-reduction reactions and the conditions of precipitation. ii) To learn about the detail's idea regarding different concepts on acid base and HSAB theory as well as the idea regarding indicators and their applications in chemical reactions. iii) To illustrate the basic concepts of 2 nd Law of thermodynamics and related state functions with the criterial concepts for spontaneity and equilibrium of the reactions. iii) To understand critically rate law of different types of reactions and dependence of rate constant on temperature, concentration etc. iv) To introduce the concept of the homogeneous catalysis and their application on the biological reactions.	4
СНЕМНР-3	Inorganic Chemistry – IB &Physical Chemistry – IB	i)Demonstrate redox reactions and teach estimation of different metals.ii) To familiarized with different orders of Chemical kinetics.	2

СНЕМНТ-4	Theory:	i)To provide students understanding on	4
	Stereochemistry,	advance stereochemistry and confirmational	
	General	analysis of different organic molecules.	
	Treatment of	ii) To learn the basic concepts of reaction	
	Reaction	energetics and understand the mechanism of	
	Mechanism,	substitution and elimination reactions.	
	Substitution and		
	Elimination		
	Reactions		
СНЕМНР-4	Organic	Preparation, purification, melting point	2
	Chemistry – II	determination and yield calculation for various	
		organic compounds to enhance the skill of	
		different methodology of organic synthesis.	

CBCS 3rd SEM honours

Course code	Course title	Course Outcome	Credit
CHEMHT-5 (Physical)	Transport processes, Applications of Thermodynamic s – I, Foundation of Quantum Mechanics,	ii) To apprise students about various application of thermodynamics such as	4
СНЕМНР-5	Physical Chemistry – II	To provide students knowledge about transport properties of solutions and electrolytes and chemical equilibria.	2

СНЕМНТ-	Theory:	i) To give the details knowledge on bonding	4
6	Chemical	mainly the ionic and covalent bonding.	-
	Bonding – I,	ii) To learn the basic knowledge of	
	Chemical	meteorology.	
	Bonding – II,	11100010108)	
	Metal extraction		
	and purification		
	from ores and		
	minerals		
	mmerais		
СНЕМНР-6	<u>Inorganic</u>	To provide the understanding on	2
	<u>Chemistry – II</u>	permanganometry and dichromatometry for	
		estimation of metals from binary mixture.	
СНЕМНТ-	Theory:	i)To give the concept about alkenes, alkynes,	4
7	Chemistry of	Carbonyl and organometallics.	
	alkenes and	ii) To familiarize with aromatic substitution	
	alkynes,	mechanism.	
	Aromatic		
	Substitution,		
	Carbonyl and		
	Related		
	Compounds,		
	Organometallics		
СНЕМНР-7	Organic	Identification and qualitative analysis of	2
	Chemistry – III	solid organic compounds.	
CHEMHS –	Basic Analytical	To enhance students skill about basic	2
1B	Chemistry	analytics of different chemical compounds.	
		Certain instrumental demonstration also	
		provided.	

CBCS 4th SEM honours

Course code	Course title	Course Outcome	Credit
CHEMHT- 8 (Physical)	Application of thermodynamics -ii, electrical properties of molecules, quantum chemistry	i)To enrich students with the knowledge of H-atom and H-like atom and solving Schrodinger equations. Also to give the basic concept about LCAO-MO and HF-SCF theory ii)To update students with phase-equilibrium chemistry and electrical properties of any molecules.	4
СНЕМНР-8	Physical Chemistry – III	i)To enable students about ionic equilibrium, potentiometric titration of different redox reactions, phase equilibrium etc. ii)Titrimetric knowledge about EMF measurements and pH values.	2
CHEMHT- 9	Radioactivity and nuclear chemistry, Chemistry of s and p-block elements, Coordination Chemistry - I	i)To introduce the concept of radioactivity and co-ordination chemistry.ii) Chemical periodicity and concept about the subsequent elements: s and p-block	4
СНЕМНР-9	Inorganic Chemistry – III	To teach students carryout different complexometric titration and to familiarized students with inorganic preparation.	2
CHEMHT- 10	Nitrogen compounds, Rearrangements, The Logic of	i)To enable students with the knowledge of N-compounds and it's rearrangement and organic spectroscopic method.ii)To understand the beginning of different C-	4

	Organic Synthesis, Organic Spectroscopy,	C bond formation and breaking required for designing synthetic route and it's viability.	
CHEMHP- 10	Organic Chemistry – IV	To demonstrate about quantitative analysis of different organic molecules through various estimation methods	2
CHEMHS- 2A	Pharmaceutical Chemistry	To enhance students skill abouts different drugs and pharmaceuticals and the basic knowledge of fermentation process has been incorporated.	2

CBCS 5th SEM honours lesson Plan

Course code	Course title	Course Outcome	Credit
СНЕМНТ-11	Coordination	i)To introduce students with the	4
	Chemistry – II,	spectrochemical series and a details idea	
	Magnetochemistr	about the crystal field theory with a	
	y, Chemistry of d-	preliminary idea about ligand field theory. To	
	and f-block	study the concept of John Teller effect and	
	elements,	it's application in Z-in and Z-out complexes.	
	Reaction Kinetics	Explain about the origin of colour and	
	and Mechanism	magnetic properties of the complexes.	
		ii) To give introductory idea about various	
		inorganic reaction mechanism through	
		concept of labile-inert complex, trans-cis	
		effect.	
		iii) Elementary concept about the d and f	
		block elements and their properties.	
СНЕМНР-11	Inorganic	i)To give idea about gravimetric estimations	2
	Chemistry – IV	and applications of permanganometry,	

		iodometry etc. ii)Inorganic preparation of complexes and it's spectrophotometric analysis.	
СНЕМНТ-12	Theory: Molecular Spectroscopy, Photochemistry, Surface phenomenon,	i)To infuse the knowledge of Rotational, Vibrational, Raman, NMR, ERS spectroscopy. ii)To develop understanding about the Law's abortion of light energy by different molecules and sequent photochemical reactions and chemical processes and it's mechanism. iii)To give a brief idea about physical and chemical adsorption and origin of stability of colloids.	4
СНЕМНР-12	Physical Chemistry – IV	To give them experiences about different spectrophotometric and surface phenomenon experiment.	2
CHEMHTDS E-1B		 i)To learn the importance of inorganic materials in industrial zone. ii)Introductory idea about renewable energy, different catalysis and chemical explosive. iii) 	4
CHEMHPDS E-1B		To give the students required expertise on different estimation processes of industrial materials.	2
CHEMHTDS E-2A	Qualitative and quantitative, Optical methods of analysis. Thermal and	i)Basic concepts of analytical chemistry and it's applications.	4

	Electroanalytica		
	l methods of		
	analysis.		
	Separation		
	techniques		
CHEMHPDS	Analytical	i)To give hands on experiences on different	2
E-2A	Methods in	separation technique such as	
	Chemistry	chromatography, solvent extraction, and ion	
		exchange methods.	
		ii)Spectrophotometric determinations of	
		indicators and soil.	

CBCS 6th SEM honours

Course code	Course title	Course Outcome	Credit
CHEMHT -13	Molecular Symmetry and Point group, (12 L) Bio-inorganic Chemistry, Organometall ic Chemistry and Catalysis	i)To give an elementary idea about application of symmetry on some simple inorganic molecules through the concept of point group. ii) Introduce the concept of bio-inorganic molecules available in nature and different biological processes. iii)To acquaint details idea about organometallic chemistry and it's application in catalysis.	4
CHEMHP -13	Inorganic Chemistry – V	To build up a concept of semi-micro qualitative analysis of different acid and basic radicals and their chemical reactions in different conditions.	2

Carbocycles	i)To develop concept in diverse chemistry of	4
and		
Heterocycles		
-	•	
Stereochemis		
try,		
Pericyclic		
reactions,		
Carbohydrate		
s,		
Carbohydrate		
S,		
Biomolecule		
S		
0	iVT	2
_		2
•		
V		
	certain compounds.	
Crystal	i)To enrich students with the knowledge of details of	4
Structure,	crystal structure and it's determination method of	
Statistical	solid compounds.	
Thermodyna	ii)To introduce concept of statistical	
mics, Special	thermodynamics and it's applications.	
selected	iii) To give knowledge about macromolecules and	
topics,	it's formation kinetics.	
Advanced	To benefits students with the knowledge of computer	2
Physical	applications in aspects of solving various numerical	
Chemistry	problems of chemistry.	
	Heterocycles , Cyclic Stereochemis try, Pericyclic reactions, Carbohydrate s, Carbohydrate s, Biomolecule s Organic Chemistry – V Crystal Structure, Statistical Thermodyna mics, Special selected topics, Advanced Physical	heterocycles, carbocycles, Cyclic Stereochemistry and natural products. ii) To give a foundation leaning in pericyclic reactions through FMO approach. reactions, Carbohydrate s, Biomolecule s Organic Chemistry — chromatographic separations of different ammino acids, dyes, sugars etc. ii) Key idea of spectroscopic analysis of organic compounds of 1H-NMR and IR spectroscopy of certain compounds. Crystal Structure, Structure, Structure, Statistical Thermodyna mics, Special selected ii) To give was thorough experiments on chromatographic separations of different ammino acids, dyes, sugars etc. ii) Key idea of spectroscopic analysis of organic compounds of 1H-NMR and IR spectroscopy of certain compounds. Crystal Structure, Structure, Structure, Structure, Structure, Structure, Structure, Structure, Structure, Structure and it's determination method of solid compounds. iii) To introduce concept of statistical thermodynamics and it's applications. iii) To give knowledge about macromolecules and topics, To benefits students with the knowledge of computer applications in aspects of solving various numerical

СНЕМНТ	Project Work	To manifest the total CBCS curriculum knowledge	4
DSE-4 and		by each student to plan and execute a specific work	2
СНЕМНР		with necessary literature survey with some	2
DSE-4		laboratory work.	
		The outcome should be presented by each student in	
		a systematic manner. To trained and inspired each	
		students to take the research work in their future	
		carrier.	

➤ Programme Course wise Credit Distribution and Outcome

Course	Total	Total	Credit					
	no of	Theor	Practi	Core	Critical	Analytical	Resea	Team
	Paper	у	cal	compete	Thinking	Reasoning	rch	Work
	S			nce			Skill	
Core	12	48	24	$\sqrt{}$	$\sqrt{}$	V	$\sqrt{}$	$\sqrt{}$
Courses								
Disciplin e Specific Elective	6	24	12	V	V	V	V	V
Skill Enhance ment	4	8	-	V	V	X	X	X

CBCS 1st SEM Programme and GE

Course code	Course title	Course Outcome	Credit
CHEMGT-1	Atomic Structure, Chemical Periodicity,	i)To give clear concepts about Atomic Structure, Chemical	4
	Acids and	Periodicity, Acids and Bases,	
	Bases,Redox	Redox Reactions, and elementary	
	Reactions, General	idea about stereochemistry,	

	Organic Chemistry & Aliphatic Hydrocarbons	substitution and elimination reactions and aliphatic hydrocarbons.	
CHEMGP-1	Inorganic Chemistry - I & Organic Chemistry – I	 i)To introduce estimation processes through Permanganometry, dichromatometry, iodometry and acid-base mixture. ii) To learn identification of single solid through Qualitative analysis. 	2

CBCS 2nd SEM Program & GE

Course code	Course title	Course Outcome	Credit
CHEMGT-	States of Matter &	i)To give a comprehensive idea about Kinetic	4
2	Chemical	theory of gases, liquids, solids and Chemical	
	Kinetics,	kinetics.	
	Chemical	ii) To give thorough idea about chemical	
	Bonding &	bonding and molecular structure and properties	
	Molecular	of p-block elements.	
	Structure, P-		
	Block		
CHEMGP-2	Physical Chemistry	i)Experiments on kinetics, viscosity, surface	2
	– I &Inorganic	tension through different apparatuses.	
	Chemistry – II	ii)Qualitative semi-micro analysis of	
		3radicals.s	

CBCS 3rd SEM Program

Course	Course title	Course Outcome	Credit
CHEMGT	Chemical	i)To give clear idea about thermodynamics,	4

-3	Energetics,	chemical equilibria and ionic equilibria.	
	Equilibria,	ii) To introduce elementary functional group	
	Organic	approach for different reactions of aromatic	
	Chemistry-	hydrocarbons, organometallic compounds, aryl	
	II	halides, alcohol, phenol, ethers and carbonyl	
		compounds.	
CHEMGP	Physical	i)To study different experiments of	2
-3	Chemistry - II	thermochemistry, ionic equilibria and solubility.	
	&Organic	ii) Identification of few pure organic	
	Chemistry - II	compounds.	

CBCS 4th SEM Program

Course	Course title	Course Outcome	Credit
code			
CHEMGT	Solutions, Phase	i)To understand the transport	4
-4	Equilibria,	property, phase equilibria,	
	Conductance,	electrochemistry and concepts of	
	Electrochemistry	solutions.	
	, Transition	ii) To introduce transition elements	
	Metal &	(3d) and it's relations with the co-	
	Coordination	ordination chemistry and crystal	
	Chemistry	field theory.	
CHEMGP	Physical	i)Hands-on experiment on	2
-4	Chemistry - III	equilibrium, conductometric and	
	&Inorganic	potentiometric titration.	
	Chemistry – III	ii)To give idea on Complexometric	
		estimation and complex synthesis	

CBCS 5th SEM Program lesson Plan

Course	Course title	Course Outcome	Credit
CHEMGT DSE-5	Analytical, Environmental and Industrial Chemistry	To give introductory idea about Analytical, Environmental and Industrial Chemistry	4
CHEMGP DSE-5	Analytical and Environmental Chemistry	Experiment on acid-base redox reaction to identify the pH range and estimation of hardness of water and strength of H_2O_2 sample.	2

CBCS 6th SEM Program lesson Plan

Course	Course title	Course Outcome	Credit
code			
CHEMGT DSE-2	Advanced Organic Chemistry and Industrial Chemistry	i)To take again the functional group approach for different organic reactions of carboxylic acids and their derivatives, diazonium salts, amino acids and carbohydrates ii)To explain the basic concepts of different chemical products from industries.	4
CHEMGP DSE-2	Advanced Organic Chemistry &Industrial Chemistry	To learn certain nitration and condensation reactions and derive yield of the crude product. To learn experiment for estimation of different commercial products	2